

FARNESYL-DERIVED INHIBITORS OF RAS FARNESYL TRANSFERASE

Mohinder S. Kang,^{1*} David M. Stemerick,¹ John H. Zwolshen,¹
Brenda S. Harry,¹ Prasad S. Sunkara,² and Boyd L. Harrison¹

¹*Hoechst Marion Roussel, Inc., 2110 E. Galbraith Rd.,
Cincinnati, Ohio 45215*

²*Neurocrine Biosciences Inc., 3050 Science Park Dr.,
San Diego, California 92121*

Received October 3, 1995

Summary: The synthesis and evaluation of farnesyl-derived inhibitors of *ras* farnesyl transferase are presented. Evaluation of inhibitors of farnesyl transferase and comparison with the previously described inhibitor was accomplished using purified enzyme and Amersham's Farnesyl:Transferase enzyme assay kit. These results show an order of magnitude increase in inhibitory activity for β -ketophosphonic acid over β -hydroxyphosphonic acid. Incorporation of fluorines in α -position, led to an increase in inhibitory activity over the nonfluorinated analogues. © 1995 Academic Press, Inc.

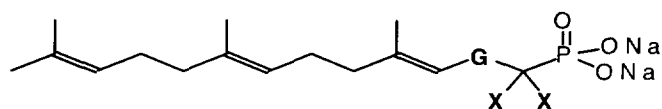
Mammalian *ras* genes encode for GTP-binding proteins involved in signal transduction pathways for cell growth and differentiation(1-3). The occurrence of *ras* oncogenes in 30% of human tumors (4) along with the elucidation of the posttranslational steps required for activity of the *ras* proteins (5) has made inhibition of *ras* protein function an emerging target for drug design for tumors in which *ras* may contribute to tumor formation (6).

Membrane localization is required for activity of the oncogenic *ras* proteins. Localization is dependent on three posttranslational modifications; farnesylation, proteolysis, and methyl esterification. Farnesylation of a specific cysteine residue near the carboxy terminus, such as Cys186 of Ha-p21ras, is catalyzed by farnesyl transferase using farnesyl pyrophosphate as the isoprenoid donor. A mechanistic study of farnesyl transferase has shown a random sequential order for the binding of substrates for this Mg^{2+} and Zn^{2+} dependent enzyme (7). Proteolytic removal of three carboxyl-terminal residues and methyl esterification of the new carboxyl-

0006-291X/95 \$12.00

Copyright © 1995 by Academic Press, Inc.
All rights of reproduction in any form reserved.

terminus farnesylated cysteine provides a protein capable of membrane localization. Interference with membrane localization by inhibition of any of these three steps may provide a means for intervention with a chemotherapeutic agent (6). The sequence requirement for peptidyl substrates of farnesyl transferase has been delineated (8) and a number of small peptide-based and peptidomimetic inhibitors of that enzyme have been prepared (9-11). Several of these peptide-derived inhibitors decrease the amount of *ras* protein processed leading to an increase in unprocessed cytosolic *ras* protein which has a negative inhibitory effect on membrane bound mutant *ras* protein (12). Proteins other than *ras* are isoprenylated *in vivo*; most are modified by a C-20 geranylgeranyl moiety by a transferase similar to farnesyl transferase. Both of these transferases have a common α -subunit, however, the β -subunit of each is distinct and determines whether the transferase will catalyze the transfer of a farnesyl or a geranylgeranyl group. A specific inhibitor of farnesyl transferase which minimized interference with other isoprenylations would be most useful. To avoid the metabolic and bioavailability problems often associated with peptide-based inhibitors, as well as interference with the function of geranylgeranyl modified proteins, we synthesized ketones **1** and **2** and alcohols **7** and **8**, as mimetics of farnesyl pyrophosphate and tested their potential as inhibitors of farnesyl transferase.



1: X = F, G = CO

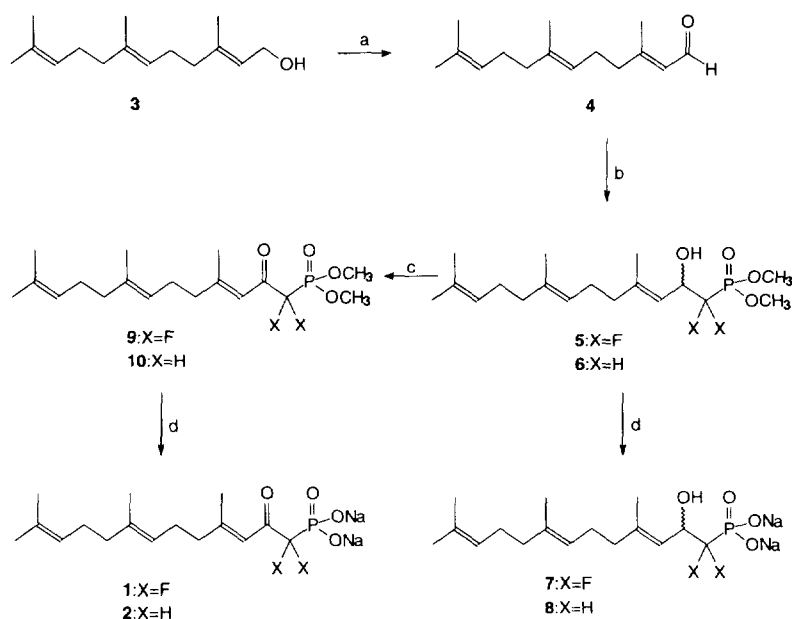
2: X = H, G = CO

7: X = F, G = CHOH

8: X = H, G = CHOH

MATERIALS AND METHODS

These inhibitors were prepared as shown in Scheme I (13). Farnesal **4** was prepared, reproducibly and in essentially pure form by oxidation of farnesol **3** with BaMnO_4 (14). The addition of lithio dimethyl methylphosphonate (15) to **3** proceeded smoothly to provide dimethyl β -hydroxyphosphonate **6**. The reaction of lithio dimethyl difluoromethylphosphonate (16) with **3**, however, required careful monitoring of the reaction temperature both during the formation of the anion and during the addition of the aldehyde. Reaction temperatures above -72°C caused an exothermic decomposition of the reagent that led to recovered starting aldehyde and greatly decreased yields of dimethyl β -hydroxyphosphonate **5**. Oxidation of **5** by the method of Swern using trifluoroacetic anhydride (17) gave dimethyl β -ketophosphonate **9**. Oxidation of dimethyl β -hydroxyphosphonate **6** with BaMnO_4 (14) provided dimethyl β -ketophosphonate **10**. The dimethyl β -ketophosphonates **9** and **10** and the dimethyl β -hydroxyphosphonates **5** and **6** were demethylated and purified according to the method of Biller (18) to give β -ketophosphonic acids **1** and **2** and β -hydroxyphosphonic acids **7** and **8**, respectively. Evaluation of compounds and comparison with previously described inhibitor **11** (7) was performed as described below.



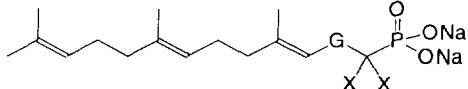
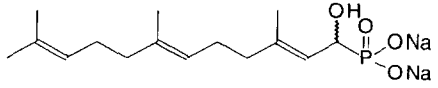
Scheme 1. Synthesis of Inhibitors

Reagents and Conditions: a) BaMnO_4 , CH_2Cl_2 ; 5 days; b) when X is H, dimethyl methylphosphonate, LDA, THF, -70°C , and when X is F, dimethyl difluoromethylphosphonate, LDA, THF, $<-72^\circ\text{C}$; c) when X is H, BaMnO_4 , CH_2Cl_2 , 5 days, and when X is F, 1) DMSO, TFAA, CH_2Cl_2 , 2) NEt_3 ; d) TMSI, collidine, CH_2Cl_2 , 0°C .

Enzyme Purification: 100 g of pig kidney (Pel-freez Biologicals, Ark.) was homogenized for 2 minutes in buffer A (100 mM Tris pH 7.5, containing 1 mM of each EDTA, EGTA, DTT and $1\mu\text{g/mL}$ each of the protease inhibitors leupeptin, antipain, pepstatin, and 0.1 mM caproic acid and 0.5 mM PMSF). The homogenate was spun at $10,000 \times g$ for 60 minutes to get the soluble fraction. The supernatant was made 50% with ammonium sulfate and stirred at 4°C for 30 minutes and spun at $12,000 \times g$ for 30 minutes to get the pellet. The pellet was dissolved in buffer B (50 mM Tris pH 7.5 containing 1 mM DTT and $20\mu\text{M}$ ZnCl_2) and dialyzed to recover the enzyme. The ammonium sulfate precipitated enzyme was further purified on a column Q (Q Sepharose, Pharmacia) and eluted with 200-800 mM NaCl gradient in buffer B. Partially purified enzyme eluted with around 500 mM NaCl was used as source of enzyme. The enzyme stored at -70°C after dialysis against buffer B is stable under these conditions for several months.

Assays: The assay is based on the scintillation proximity principle. A human lamin-B carboxyl-terminus sequence peptide linked to biotin is ^3H -farnesylated at the cysteine residue by farnesyl:transferase. The resultant complex is captured by a streptavidin-linked SPA bead. In a $100\mu\text{L}$ total reaction volume, the test compound was added in $10\mu\text{L}$ of solvent, usually DMSO, then $70\mu\text{L}$ of the reaction mixture containing $10\mu\text{L}$ of 1% Triton X-100, $20\mu\text{L}$ of 1:100 diluted ^3H -farnesylidiphosphate (0.2 μCi), $20\mu\text{L}$ of assay buffer (50 mM HEPES, 30 mM MgCl_2 , 20 mM KCl, 5 mM DTT, and 0.01% Triton X-100), and $20\mu\text{L}$ of 0.5 μM Biotin-Lamin B peptide in buffer pH 7.5 (50 mM HEPES, 25 mM Na_2HPO_4 , 20 mM KCl, 5 mM DTT, and 0.01% Triton X-100) was added. This mixture was allowed to stand at 37°C for 5 minutes. The reaction was started by adding of $20\mu\text{L}$ of enzyme ($\sim 3\mu\text{g}$ of protein). The mixture was incubated at 37°C for 30 minutes and the reaction was stopped by adding $150\mu\text{L}$ of the stop/bead reagent. The samples were counted in a Packard Tri-Carb 1900 CA scintillation counter.

Table I. Inhibition of Farnesyl Transferase.

			Inhibition of Farnesyl Transferase
compound	G	X	IC ₅₀ (μM)
1	-C(O)-	F	0.35
2	-C(O)-	H	1.56
7	-CH(OH)-	F	0.78
8	-CH(OH)-	H	11.5
			0.15
11			

RESULTS AND DISCUSSION

The inhibitory values given in Table I suggest that all of the farnesyl phosphonic acid derivatives prepared were good inhibitors of farnesyl:transferase. These results show almost an order of magnitude increase in inhibitory activity for β -ketophosphonic acid **2** over β -hydroxyphosphonic acid **8**. Incorporation of fluorines at the α -position, i.e. β -ketophosphonic acid **1** and β -hydroxyphosphonic acid **7**, led to a 5-15 fold increase in inhibition over the non-fluorinated analogues. It can be seen that ketone **1**, the most electrophilic ketone, is the most potent inhibitor showing comparable inhibition to the literature standard **11**. The next most potent inhibitor is α,α -difluoro alcohol **7**. These facts are most consistent with the pK_a of the phosphonate moiety controlling the inhibitory activities of the compounds prepared. α,α -Difluoro substitution is well known to increase the acidity of phosphonate group and, thus more closely mimic a phosphate moiety. These results further suggest that farnesyl derived inhibitors may have potential as chemotherapeutic agents.

REFERENCES

1. Maltese, W. A. (1990) The FASEB Journal 4, 3319-3328.
2. Graves, J. A.; Donward, J.; Izquierdo-Pastor, M.; Rayter, S.; Warne, P. H.; Cantrell, D. A. (1992) J. Immunology 148, 2417-2422.
3. Cox, A. D.; Hisaka, M. H.; Buss, J. E.; Der, C. J. (1992) Molecular and Cellular Biology 12, 2606-2615.
4. Bos, J. L. (1989) Cancer Research 49, 4681-4689.
5. Casey, P. J.; Solski, P. A.; Der, C. J.; Buss, J. E. (1989) Proc. Natl. Acad. Sci. USA 86, 8323-8327.
6. Gibbs, J. B. (1991) Cell 65, 1-4.

7. Pompliano, D. L.; Rands, E.; Schaber, M. D.; Mosser, S. D.; Anthony, N. J.; Gibbs, J. B. (1992) *Biochemistry* 31, 3800-3807.
8. Reiss, Y.; Stradley, S. J.; Gierasch, L. M.; Brown, M. S.; Goldstein, J. L. (1991) *Proc. Natl. Acad. Sci. USA* 88, 732-736.
9. Goldstein, J. L.; Brown, M. S.; Stradley, S. J.; Reiss, Y.; Gierasch, L. M. (1991) *J. Biol. Chem.* 266, 15575-15578.
10. Graham, S. L.; deSolms, S.J.; Giuliani, E. A.; Kohl, N. E.; Mosser, S. D.; Oliff, A.I.; Pompliano, D. L.; Rands, E.; Breslin, M. J.; Deana, A. A.; Garsky, V. M.; Scholz, T. H.; Gibbs, J. B. and Smith, R. L. (1994) *J. Med. chem.* 37, 725-732.
11. James, G.L.; Goldstein, J.L.; Brown, M.S.; Rawson, T. E.; Somers, T.C.; McDowell, R. S.; Crowley, C. W.; Lucas, B. K.; Levinson, A. D. and Masters, J. C. Jr. (1993) *Science* 260, 1937-1942.
12. Gibbs, J. B.; Schaber, M. D.; Schofield, T. L.; Scolnick, E. M.; Sigal, I. S. (1989) *Proc. Natl. Acad. Sci. USA* 86, 6630-6634.
13. All new compounds gave satisfactory elemental analysis and IR, mass, ^1H , ^{19}F , ^{31}P NMR spectral data consistent with the assigned structure.
14. Firouzabadi, H.; Ghaderi, E. (1978) *Tet. Let.* 838-840.
15. Marquez, V. E.; Lim, M-I; Tseng, C. K. H.; Markovac, A.; Preist, M. A.; Khan, M. S.; Kaskar, B. J. (1988) *J. Org. Chem.* 53, 5709-5716.
16. Obayashi, M.; Ito, E.; Marsui, K.; Kondo, K. (1982) *Tet.Let.* 23, 2323-2326.
17. Mancuso, A. J.; Swern, D. (1981) *Synthesis* 165-185.
18. Biller, S. A.; Forster, C. (1990) *Tetradedron* 46, 6645-6658.